

Cross-Validation of a Mobile N₂O Flux Prototype (IPNOA) Using Micrometeorological and Chamber Methods

Patricia Laville¹, Simone Neri², David Continanza², Luca Ferrante Vero², Simona Bosco³ and Giorgio Virgili²

1. EGC (Environnement et Grandes Cultures) Joint Research Unit, National Institute for Agriculture Research, Thiverval-Grignon 78850, France

2. West Systems L.L.C. (Limited Liability Company), Firenze 50126, Italy

3. Institute of Life Sciences, Advanced School Sant'Anna, Pisa 56127, Italy

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Abstract: Within a LIFE+ project IPNOA (improved flux prototype for n₂o emission from agriculture), LIFE11 ENV/IT/302 is a mobile prototype was developed to evaluate at field scale N₂O emissions using a fast chamber technique. Main challenge was to develop a mobile system capable of moving on various field surfaces, equipped with very reliable N₂O gas analyser (Los Gatos Research Inc.), electrically autonomous (with batteries) and enough robust to face up to field conditions. In this paper, we report the major features of this prototype studied during two field campaigns. The N₂O flux IPNOA prototype was compared with other methodological implementations: first, during an INGOS (integrated non-CO₂ greenhouse gas observing systems) campaign on a grazed grassland at Easter Bush (Scotland) by Eddy correlation method, and then after on an arable crop at Grignon (France) using automatic and manual chambers fitted with QC-TILDAS (Quantum Cascade Tunable Infrared Laser Differential Absorption Spectrometer, Aerodyne Research Inc.), with the 46C model of thermo-instrument analyser or with a GC (gas chromatography) analysis.

Key words: Monitoring device, nitrous oxide inventory, agricultural managements, steady state chamber.

1. Introduction

In order to implement the European strategy on the reduction of greenhouse gas emissions of 20%-30% in 2020, an important decrease in the output of all these gases from their main sources is needed.

Agriculture is responsible in Italy for about 6% of total GHG (greenhouse gas) emissions in 2014 [1], with a main contribution by nitrous oxide (N₂O) gas release from soil (i.e., 73%). Nitrous oxide emissions are mainly due to nitrogen fertilization in cropping systems. Nowadays, the total amount in N₂O emissions for agricultural soil are mainly estimated by using the IPCC (Intergovernmental Panel on Climate Change) guidelines, reporting a simplified model involving the amount of fertilizer applied only and leaving out of

consideration other potentially important parameters [2], such as climatic conditions soil properties and agricultural managements.

N₂O emissions are very difficult to evaluate, insofar magnitude of emissions are very variable spatially and temporally at field scale and all through the year.

The most common technique for assessing N₂O fluxes on many crop plots at once is the steady state chamber [3, 4]. For the steady state chambers, N₂O concentration changes are the most frequently estimated by both taking gas samples in small vacuum vials at different moments after closing the chamber and analyzing them thereafter using GC (gas chromatography) methodology. This involves many manipulations: making the vacuum in the vials, collecting the samples in the field after closing the chamber, ensuring the supply of carrier gas for the GC analysis, identifying and placing the tubes on an auto

Corresponding author: Patricia Laville, doctor, research field: environmental sciences. Email: laville@grignon.inra.fr.

sampler rack, performing calibration ranges for samples, formatting analysis. In addition, for each sample, the analysis lasts few minutes and limits the number of samples. The sensitivity of standard GC analysis is also weak, consequently to detect low emissions, it is necessary to maintain chamber closed during large period of few tens of minutes (30-60 min) and can disturb the diffusion of biogas from the soil to the surface.

Developments of methodologies to improve inventories of these emissions are still needful. The LIFE+ IPNOA project fits perfectly within this requirement. Indeed, major objectives of this project were the developments of two prototypes for measuring N₂O fluxes, the first “portable” to evaluate spatial variation on N₂O emission at field scale for various crops and managements, the second “fixed” (the “station” prototype) to obtain the seasonal variations in emission.

This paper concerns only the field validation of the “portable” prototype. It was designed by WS (West System L.l.c.). To overcome the major inconvenient of steady chamber technique listed above, the prototype was equipped with on line continuous N₂O analyzers, the sensitivity of the selected analyzer was enough to allow flux measurements over periods of 5 min. Compared to fast box system [5], the prototype was most “mobile” allowing measurements all across a field of several hectares or for contiguous fields. The prototype was designed to be independent in 230 VAC (volts of alternative current) power supply or generator.

The objective of this paper was to present the major characteristics of the IPNOA prototype and the results of a cross validation with various other methodologies previously used by INRA-EGC (Environnement et Grande Culture Joint Research Unit, National Institute for Agriculture Research). This validation was achieved during two experiments. The first was conducted in Scotland (Edinburgh) grassland during an INGOS (integrated non-CO₂ greenhouse gas observing systems) campaign. The mapping of N₂O emissions

obtained with the mobile prototype was compared with measurements of N₂O fluxes using EC (Eddy covariance) methodology. The second experiment was conducted at Grignon (France) on barley crop. Performances of the prototype were then compared with other chamber designs (manual and automatic chambers). These campaigns were also an opportunity to test various features of the prototype.

2. Materials and Methods

2.1 Description of the Mobile Prototype

The instrumentation is arranged on a detachable unit placed on the light vehicle in order to separate the part dedicated to the traction from the part dedicated to the measurement. The vehicle itself hosts the batteries and the traction engines and in the upper part the control unit with gas analyzers (Fig. 1). There is two packs of batteries on dedicated to the vehicle engines (powered with 24 VDC (volts of direct current)), the other for the instrumentation.

The vehicle is equipped with remote motion controller. A joystick allows the control of the motion in all direction. The instrumentation is placed on platform fixed on the vehicle with silent block in order to dampen the vibration during the transport. The platform hosts two gas analyzers: N₂O/CO (nitrous oxide, carbone monoxide) and UGGA (ultraportable greenhouse gas analyzer) analyzers developed by LGR (Los Gatos Research Inc.) [6]. These spectrometers use LGR’s patented off-axis ICOS (integrated cavity

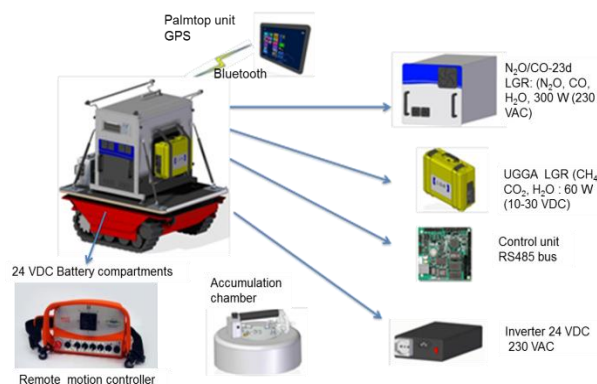


Fig. 1 Light vehicle and detachable instrumentation.

output spectroscopy) technology, a fourth generation cavity (<http://www.lgrinc.com/analyzers/>). The first one measures the N₂O, CO and H₂O gases and requires 300 W (powered with 230 VAC) in operation, the second measures CH₄, CO₂ and H₂O and requires 60 W (powered with 10-30 VDC). AC (alternating current) power of 1,500 W for the instrumentation supply is delivered by an inverter of 24 VDC input from the battery pack. The battery packs are recharged by a battery charger box that will be connected to the power grid (230 VAC), when the prototype is not deployed in the field. The gas analyzers have several interfaces USB (universal serial bus), SERIAL (RS-232 standard), Ethernet ports. Output gas concentrations are given with a scan rate of 1 s.

A control unit operates/interrogates the various outputs of the instrumentation. The control unit is connected through bluetooth to a palmtop unit. The palmtop unit acquires and stores the measured values. The palmtop unit is equipped with a high performance GPS (global positioning system) allowing locating the chamber positions during its deployment.

The two gas analyzers are connected in series according to Fig. 2. The LGR N₂O has an internal pump (flow rate of about 250 cc min⁻¹). The sample inlet and outlet are 1/4 swagelok fittings. These ports will be connected to 2-4 mm (ID/OD (internal /external diameters)) teflon (or rilsan) tube. Twenty meters of tubes separate the inlet/outlet gas analyzers to the inlet/outlet accumulation chambers. Several chamber designs were tested during our cross validation (Fig. 3). All the chambers are equipped with a fan that mixes the headspace air during the measurement. The ‘C’ chamber (V/A = 0.104 m) was the most used.

The entire device weighs about 600 kg and can be transported from on site to another at the rear of a van.

A software “flux manager” installed on the palm pilot unit manages the start and the stop of the data acquisition during the chamber deployment. It is very interactive software, in real time, it is possible to

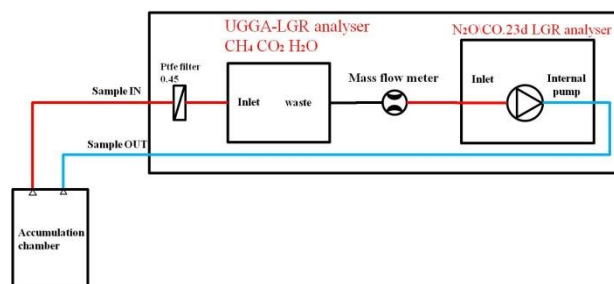


Fig. 2 Configuration of the IPNOA prototype circuit for the gases the LGR UGGA analyzer and N₂O LGR analyzers are in series.



Fig. 3 Design of the IPNOA chambers.

A: area = 3.33 dm², volume = 2.9 dm³, V/A = 0.087 m;
 B: area = 3.40 dm², volume = 6.2 dm³, V/A = 0.211 m;
 C: area = 7.35 dm², volume = 6.65 dm³, V/A = 0.140 m.

adjust the time interval to evaluate the rate of gas accumulation.

Gas emissions are deduced with concentration increase during the chamber deployments according a linear model $C_i = bt_i + a$. C_i denotes to the measured concentration at time t_i , b is the slope coefficient and a is the intercept at $t_i = 0$. Emissions are proportional of the b coefficient and are estimated according to the following relationships:

$$Flux = \frac{V}{A} b \quad (1)$$

The V/A corresponds to the ratio between the headspace volume and the area of the chamber.

$$b = \frac{\sum_{i=1}^n (C_i - \bar{C}) (t_i - \bar{t})}{\sum_{i=1}^n (t_i - \bar{t})^2} \quad (2)$$

A post treatment was performed according to two criteria. It must maximise the slope magnitude and the correlation coefficient, R^2 .

Two criteria were used to evaluate the quality of the

measurements: the coefficient of determination R^2 and the standard deviation of the slope coefficient— Δb using also to evaluate the standard deviation of the fluxes ($Flux$). Their analytical equations are the following:

$$R^2 = 1 - \frac{\sum_{i=1}^n (C_i - bt_i - a)^2}{\sum_{i=1}^n (C_i - \bar{C})^2} \quad (3)$$

$$\Delta Flux = \frac{V}{A} \Delta b \quad (4)$$

With:

$$\Delta b = \sqrt{\frac{1}{n-2} \frac{\sum_{i=1}^n (C_i - bt_i - a)^2}{\sum_{i=1}^n (t_i - \bar{t})^2}} \quad (5)$$

When the linear model $C_i = bt_i + a$ fits well the data, R^2 is close to 1 and the $\Delta b/b$ ratio should be as small as possible.

2.2 Scotland Green Grass Experimentation: “Easter Bush” Site

The “Easter Bush” campaign took place during a N₂O inter-comparison campaign of INGOS project. INGOS is an EU funded IA (integrating activity) project targeted at improving and extending the European observation capacity for non-CO₂ greenhouse gases (<http://www.ingos-infrastructure.eu/>).

To support this activity, the CEH (Centre for Ecology and Hydrology) in Edinburgh (UK) hosted an inter-comparison campaign for the measurement of micrometeorological fluxes of N₂O on managed grassland for the period 3rd to 28th June, 2013. Measurements with IPNOA prototype were achieved during the last week of June 2013.

The Easter Bush measurement site is located in a rural area 10 km south of Edinburgh, Scotland UK (3°12' W, 55°52' N, 190 m a.s.l. (above sea level)). The site is situated on the border between two intensively-managed grassland fields of approximately 5 ha (hectare) each (Fig. 4).

The organic rich soil has consistently resulted in

high N₂O emission factors from fertiliser application in previous studies. On June 11, 50 kg of nitrogen as NH₄NO₃ fertilizer form was applied. Eddy covariance methodology was mainly employed during this campaign. To inter-compare the instrumentations, the gas analysers belonging to the various participants were connected on a single EC acquisition chain (Table 1). Comparisons between instruments were very satisfactory and similar emissions were estimated with the various analyzers. Then, we compared only EGC-INRA measurements obtained by EC methodology with the dynamic enclosure method of IPNOA prototype. It was interesting to compare an “integrative” EC method with local measurements obtained with the mobile prototype.

The IPNOA chamber was deployed on the entire



Fig. 4 Easter Bush site.

Table 1 Instrumentation of Ester Bush campaign partners.

Country	Institute	Instrument	Method
Italy	West system Srl	N ₂ O-LGR	Dynamic enclosure
France	EGC-INRA	Aerodyne-QC -TILDAS*	Eddy covariance
Finland	FMI (Finnish Meteorological Institute)	N ₂ O-LGR	Eddy covariance
Finland	University Helsinki	N ₂ O-LGR	Eddy covariance
Denmark	Technical University of Denmark	Aerodyne-QC -TILDAS	Eddy covariance
Netherlands	ECN (Energy Research Centre Netherlands)	Aerodyne-QC -TILDAS	Eddy covariance
United Kingdom	CEH	Aerodyne-QC -TILDAS	Eddy covariance

QC-TILDAS*—quantum cascade tunable infrared laser differential absorption spectrometer.

surface corresponding to the fetch of Eddy covariance measurements allowing evaluating the spatial heterogeneity in emissions.

Fig. 5 shows, for example, the localizations of IPNOA samplings during Easter Bush campaign on June 26.

2.3 Grignon Barley Experimentation

The cross validation at Grignon took place from the 11th to 12th July. The cross validation was provided in a field of 1 hectare close the EGC INRA building (Fig. 6) (48°50' N 1°56' E, 127 m a.s.l.). This field was sown early on March 2013 with spring barley. One hundred and twenty N kg ha⁻¹ as NO₃NH₄ granule were applied early April. In July, the barley crop was very developed and we were away of the nitrogen application. To ensure good conditions for the cross validation, we cut barley and brought different doses of nitrogen as ammonium nitrate on different plots used as test areas.

Various implementations and equipment's were used during the cross validation: three automatically and three manually chambers were deployed on the field (Fig. 7). Performances of LGR N₂O spectrometer

of the mobile prototype were compared with other INRA gas analyzer: (a) a filter correlation spectrometer (46C Thermo Environment Inc.); (b) a GC (Varian CP3000), four certified cylinders of N₂O calibration gases supplied by Air Liquid company were also used to test sensitivity in response of the LGR N₂O analyzer and the response time of the of IPNOA chambers.

Three doses of nitrogen as ammo-nitrate granules were applied on the three manual and automatically chambers (i.e., 0 N: eq. 0 kg N ha⁻¹, 50 N: eq. 50 kg N ha⁻¹, 200 N: eq. 200 kg N ha⁻¹) with an application of water for an equivalent of 20 mm of rainfall July 3 in the morning. A second irrigation was also achieved July 8 in the evening. The INRA automatically chamber setup were largely using during the NitroEurope project [7]. Gas analysis was achieved on line using a gas filter correlation N₂O spectrometer (46C model of thermo environment). The sizes of chamber frames were 0.7 × 0.7 m² and were pressed into the soil at 7 cm, chamber height was 0.20 m. Commutation of a chamber to the next was achieved sequentially using various electro-valves. Deployment duration per chamber was 15 min with a complete cycle every hour. The scan rate for N₂O concentration measurement was 10 s. During the cross validation, the Thermo 46C analyser was replaced by the N₂O LGR to compare the responses of the two gas analysers on same soil area.

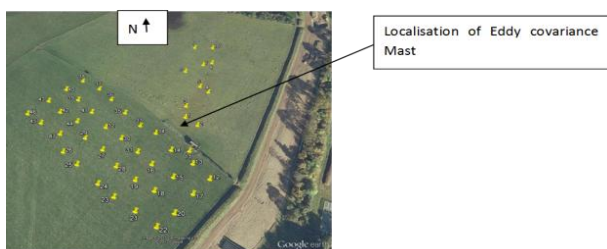


Fig. 5 Localization of the samples during Easter Bush campaign on June 26.

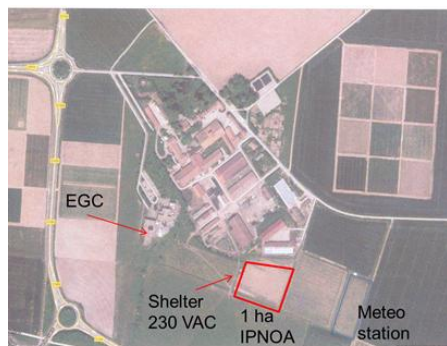


Fig. 6 Experimental site of Grignon.



Fig. 7 Grignon devices with three manual chambers on left (area = 25 dm²) and three automatic chambers on right (area = 49 dm²). The height of chamber headspace was at mean 22.6 cm for the Auto chambers and 19.4 cm for the Manual chambers. INRA chambers differ on IPNOA chamber by their volumes and areas more wide. Their frames are inserted to a deep of about 7 cm into the soil.

The size of the frames for the three manual chambers was $0.5 \times 0.5 \text{ m}^2$. They were pressed to 5 cm deep, the chamber height was also 0.20 m. Samplings of gas were collected during 45 min in vacuum tubes (12 mL) through the septum of the chamber lids every 15 min (four samples per flux determination). The air samples were then subsequently analysis by GC. On lid was specifically adapted to connect during the deployment of the chamber, the input and the output of the LGR analyzers.

3. Results and Discussions

3.1 Comparison with Eddy Covariance Methodology at Ester Bush

Extensive measurements with IPNOA mobile prototype were made on large grill of 1 ha with more than 30 localization points on June 24, 25 and 26, 2013 (Fig. 6) to determine spatial variability of N₂O emissions. IPNOA chamber measurements were compared with Eddy covariance measurements during the period from June 24 to 26 (Fig. 8). Overall, identical magnitudes of fluxes were observed between the two methodologies: emissions tend to decrease between days 24, 25 and 26. Large CVs (coefficients of variation: ratio between standard deviation to mean fluxes) were obtained (e.g., around 150% on June 26) with IPNOA chambers and indicated a large spatial variability in N₂O fluxes. In comparison, CVs were lower for CO₂ fluxes only around 30%. These larges CVs are consistent with numerous previously N₂O flux studies [7]. In the soils, N₂O production is located in “hot spots” where conditions for denitrification are favorable, i.e., anoxic zones where there is an accumulation of carbon and nitrogen as substrates for denitrifiants.

3.2 Comparison with Grignon Automatically Chambers

Fig. 9 shows the comparison between N₂O fluxes measured with Thermo 46C and the IPNOA LGR and obtained for the three auto-chamber treatments. The magnitudes of emissions obtained continuously with

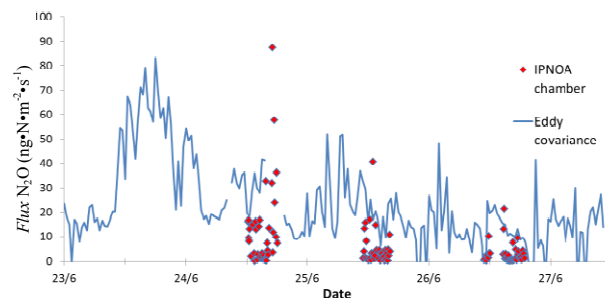


Fig. 8 Comparison of N₂O fluxes between Eddy covariance measurements and IPNOA mobile methodology. During the comparison, the wind was predominantly northerly, the footprint of the measurements corresponded to the north grassland field.

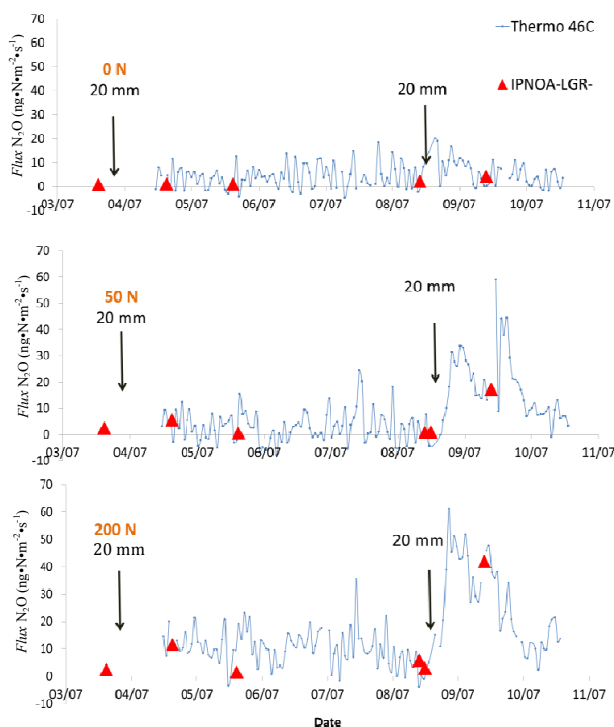


Fig. 9 Measurements with the three auto-chambers. The thermo instrument was replaced during certain sequences by IPNO analyzer (LGR) or by QCL spectrometer. The arrows indicate the nitrogen and water supplies. Only water was supplied for the second time.

the Thermo 46C instrument were weak with a mean value of 3.9, 7.5 and 14.5 $\text{ng}\cdot\text{N}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ for the 0 N, 50 N and 200 N treatments, respectively. Fluxes measured with the Thermo 46C instrument were very fluctuating. It was due to the weak precision of the spectrometer (around 12 ppb) leading to N₂O flux sensitivity around 3.4 $\text{ng}\cdot\text{N}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ for a V/A ratio of 0.2 m and an integrated time for the slope calculation of 800 s (80 samples for a scan rate of 10 s). In comparison, for the

N₂O LGR spectrometer, the ratio $\Delta b/b$ was only about 0.4% (i.e., $Flux = 0.06 \text{ ng N m}^{-2} \text{ s}^{-1}$ for a N₂O flux of $15 \text{ ng N m}^{-2} \text{ s}^{-1}$). Emissions after the first nitrogen and water application (3 July) were weak. More substantial N₂O emissions were observed only after the second irrigation, July 8 (up to $70 \text{ ng N m}^{-2} \text{ s}^{-1}$). This underlines the interest to measure N₂O fluxes with a small temporal resolution to not miss high emissions, in this case, emissions were large only after the second irrigation on July 8. Emissions were controlled by nitrogen concentration but only above a threshold in soil moisture.

For the same periods for two analyzers: LGR and Thermo 46 analyzers N₂O flux magnitudes were similar. Emissions were in relation with treatment levels in respect to the nitrogen doses with the order $0 \text{ N} < 50 \text{ N} < 200 \text{ N}$. This ranking was most distinct after the second irrigation.

3.3 Comparison with Manual Chambers

Fig. 10 shows the comparison between fluxes measured with the three manual chambers using the gas measurement methodologies: the GC, the IPNOA LGR spectrometer. For the GC, the lid of the chamber was deployed during 45 min and samples were taken every 15 min, for the LGR, it was only deployed during 4 min. For the GC measurements, mean ratio between slope and its standard deviation ($\Delta b/b$) was about 12% for 4 samples. For the GC measurements, the repeatability of the analysis leads to minimum flux detectable close to $1 \text{ ng N m}^{-2} \text{ s}^{-1}$. For the IPNOA-LGR measurements, its relative uncertainty was only 0.4%. Similar fluxes were observed between the two methods of gas analysis even if the chamber deployment durations were very different (45 min for GC analysis and 4 min for the IPNOA-LGR).

Surprisingly, fluxes for the 50 N treatment were higher than for the 200 N treatment. Nevertheless, measurements were carried out before the second irrigation July 8. Before this date, soilmoisture content was probably insufficient to discriminate N₂O production according to the level of nitrogen treatment.

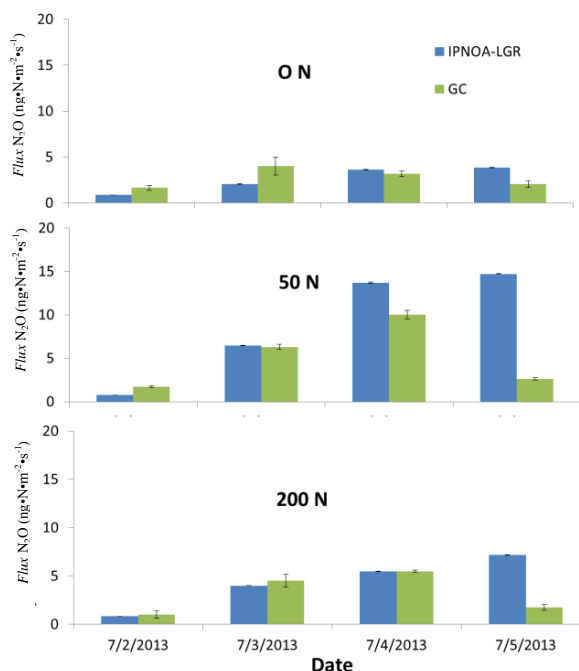


Fig. 10 Comparison between GC and LGR measurements using the frame of the three manual chambers. A lid was specifically adapted to fit the inlet and outlet of the two infra red spectrometers: LGR and QCL.

3.4 Tests with Calibration Gases: Sensitivity Response and Response Time of the Prototype

Fig. 11 shows the N₂O concentrations obtained with LGR analyzer for the various calibrated gases. LRG response with the five calibration gases shows a good sensibility with a response of only 2% lower than certified gases. The 2% correspond to the gas concentration uncertainty given by the air liquid, the supplier of the calibrated gases.

Calibrated gas cylinders were used to estimate the response time (τ_r) of the IPNOA prototype. Response time of the IPNOA prototype could be estimated with a sudden change of concentration in the inlet of the circuit.

Fig. 12 shows the variations of LGR concentrations just after connecting the input of the analyzer on a bag containing standard gas with a N₂O concentration of $0.805 \pm 0.016 \text{ ppm}$ and CO₂ of $2961.8 \pm 8.9 \text{ ppm}$. The equilibrium concentration (C_{eq}) was reached for the LGR after 100 s for CO₂ ($C_{eq} = 2,789 \text{ ppm}$) whereas for N₂O, it was reached after 200 s ($C_{eq} = 778.6 \text{ ppb}$). For N₂O measurements, we can consider that 98% of

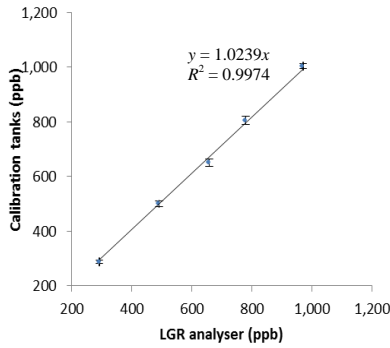


Fig. 11 Relation between LRG N₂O concentrations and calibrate gas concentrations.

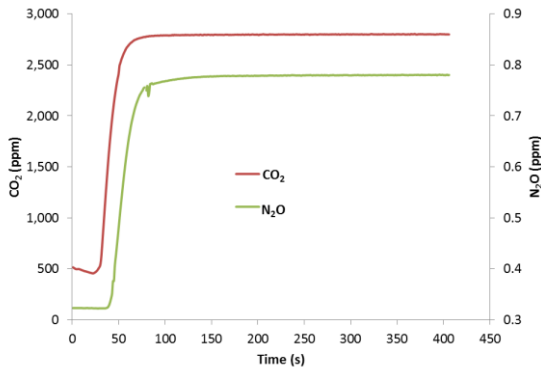


Fig. 12 Response IPNOA prototypes after connection with air-bag filled with calibrated gas.

balance is reached after 90 s, while for CO₂, this value is reached 65 s after change of concentration.

The response time determines the lag-time to consider before starting calculation of the accumulation slope for flux estimation. This time was found to be dependent mainly to the pump flow rate, the cell volumes of the LGR analyzers and length of inlet tubes. It depends only indirectly to chamber geometry (volume, area), to the extent that threshold of gas detection depend to the chamber V/A ratio.

The response time could be also estimated at first as the time necessary to purge the dead volume of the circuit considering (Fig. 2), the measurement circuit of the two analyzers in series used for the IPNOA mobile prototype. The CO₂ analyzer (UGGA) is the first on the line and the N₂O the second. The flow rate at atmospheric pressure (760 mm Hg) in the circuit is 250 cc min⁻¹. The length of the sample tubes is 20 m with an inner diameter of 2.5 mm ID (total volume about 113 cc). Cell volume for UGGA LRG was 315 cc and 401

cc for the N₂O LGR. In the cells, pressure is limited to 90 mm Hg so the flow rate is then equal to $\frac{250 \times 760}{90}$ cc min⁻¹. The response times are then equal to:

$$\tau_{CO_2} = \frac{(113)}{250} + \frac{(90 \times 315)}{250 \times 760} = 36 \text{ s}$$

$$\tau_{N_2O} = \frac{(113+16)}{250} + \frac{(316+401) \times 90}{250 \times 760} = 52 \text{ s}$$

They are close to the values estimated in Fig. 12. To measure correctly the N₂O emission, we must wait about 1 min 30 s (98% of the signal) before validating the start of accumulation phase of the N₂O, for CO₂ 1 min is enough. The TD (time duration) for chamber deployment was evaluated to 4 min 90 s for the response time and 150 s was enough to evaluate the slope regression. This time allows optimizing the maximum of the slope with a weak perturbation of the gas transfer and to maximize the R^2 .

3.5 Test of Water Dilution

The N₂O LGR also measures the water vapor concentration, so it was possible to test the effect of water dilution on N₂O flux estimations. According to Rochette and Hutchinson [4], any increase in concentration of other gases resulting from chamber placement can affect N₂O concentration. For example, Parkin and Ventura [8] demonstrate how an increase in water vapor concentration could decrease N₂O concentration by 3%. This is known as the water vapor dilution effect and may cause an underestimation of N₂O fluxes.

The raw N₂O gas concentration is given in ppb unit corresponding to the ratio between partial pressure in N₂O and total atmospheric pressure. The pressure in the LGR cell measured with barometric transducer depends also the partial pressure of water vapor (P_{H_2O}). We can write:

$$N_2O \text{ raw} = \frac{P_{N_2O}}{P_T} \quad (5)$$

N₂O raw is the pressure fraction between the N₂O gas (P_{N_2O}) and total atmospheric gases (P_T). To

overcome the variation of the total atmospheric pressure during the chamber deployment time we must consider a dry concentration in N₂O corresponding to the mixing ratio of the gas with a reference atmosphere without water vapor:

$$N_2O_{dry} = \frac{P_{N_2O}}{P_T - P_{H_2O}} \quad (6)$$

N₂O dry could be also written as:

$$N_2O_{dry} = \frac{N_2O_{raw}}{1 - H_2O_{raw}} \quad (7)$$

Fig. 13 shows an example of gas concentration accumulations after closing the chamber with and without correction on water vapor. N₂O dry concentrations are about 7 ppb higher compare to the N₂O raw concentrations. For a same measurement period, for this example, fluxes derived from the variations of the N₂O dry signal were about 6.7% higher than those observed without correction for water vapor (N₂O raw).

The effect of water vapor on N₂O flux estimations will be largely dependent to the air humidity conditions and to the evaporation/condensation at the soil surface during chamber deployments. This effect is of the order of a few percent and should not be neglected.

3.5 Test of Sealing

To properly measuring the emission by enclosure chamber method, it is important to seal tightly chamber with the soil surface. Chamber systems are often composed of a collar and a lid. The collar is pressed into the soil at few cm few days before the measurement. The lid is then fixed to the collar air tightly. During the Easter Bush and Grignon experiments, the IPNOA chamber was maintain by foot pressure to the soil surface (Fig. 14).

But the seal was found insufficient. The alternative was to use collars pressed into the soil at few centimeter and positioned at different plot on the field, the chamber is now fitted at its base with a rubber with a groove, and is clipped to collars previously deployed in the field.

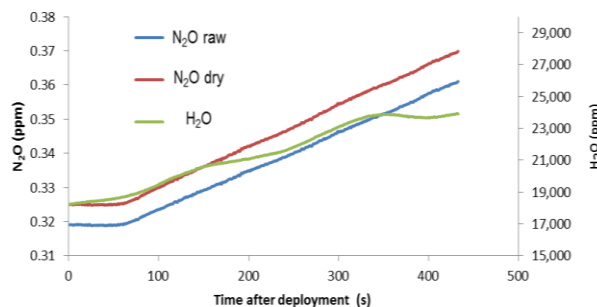


Fig. 13 Example of variation of N₂O concentration with (N₂O dry) and without (N₂O raw) correction on water vapor (H₂O).



Fig. 14 Technique initially used to compress the chamber to the soil surface.

3.6 Tests of Detection Limits

The detection limit in N₂O emissions with IPNOA system was estimated when the deployments of the chamber were correct, without any air leak and when the emissions were weak. This detection limit depends on the LGR gas analyser noise level, the integrating time duration of the deployments, and the V/A ratios of the chamber. For the chamber ‘C’ (V/A = 0.104 m) and for an integrating time of 150 s, according to the flux Eq. (4), the detection limit was around 0.03 ng N m⁻² s⁻¹. It is very nice score, with higher chamber like “B” chamber (V/A = 0.211 m), this threshold detection would have been two times higher.

The precision for ambient condition (i.e., N₂O mixing ratio around 325 ppb) given by Los Gatos Research Inc. is 0.05 ppb for a measurement rate of 1 s.

Applying the relation $\Delta Flux = \frac{V}{A} \frac{\Delta C}{\Delta t}$ with a C = 0.05 ppb and t = 150 s, we obtain same value around 0.04 ng N m⁻² s⁻¹.

It should be noted that, the precision of 0.05 ppb is given only for the ambient concentration (325 ppb). For higher concentrations, the noise level will be largest and will conduce to greatest uncertainties for large N₂O emissions.

The *b* term is inversely proportional to the root square of the measurement number. To minimize the uncertainty, it is important to establish an accumulation dynamic over more than 1 min period but after 3 min with a scan rate of 1 s, the gain on the flux accuracy is not so significant.

4. Conclusions

This cross validation demonstrated the proper functioning of the mobile IPNOA prototype according to various criteria that were tested:

- Sensitivity: the responses of the prototype according to the entire range of emissions observable in the ecosystems of a grass land and an arable crops were very satisfactory given the comparisons with other devices (Eddy covariance, auto or manual chambers), and gas analyzers (Thermo 46C, GC, Aerodyne QCL) and gas standards that were deployed during the cross validation;

- Short response time allowing limiting deployment time of the chamber. 90 s were enough to start gas accumulation phase. And 150 s to evaluate the rate of gas accumulation;

- It is operational capacities in terms of mobility with stability in response of the gas analyzer and autonomy in electric power for less than 8 h;

- The ease-of-use of the instrument: the remote transmission for the operation of the commands and acquisition of the data with a palm pilot was very use-friendly;

- The weak limit of detection was also an important specificity. It is noteworthy, around 0.04 ng N m⁻² s⁻¹ and it is largely linked to characteristics of the LGR N₂O analyzer. Five minutes of chamber deployment were enough to estimate fluxes with a high resolution;

- The main problem observed during these

experiments was related to chamber sealing during their deployments. It was overcoming using collars pressed into the soil on which was clipped chamber during its deployment.

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